



# Chem 109 C

## Bioorganic Compounds

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## **Protein Structure and Analysis (Ch 21)**

**structure: primary, secondary, tertiary, quaternary  
parameters of alpha-helix, beta sheet, loops/coils  
and tertiary structures**

**structure determination, reagents:**

**HSCH<sub>2</sub>CH<sub>2</sub>OH**

**6M HCl**

**Edman's reagent**

**cyanogen bromide BrCN**

**exopeptidases:**

**carboxypeptidase A**

**carboxypeptidase B**

**endopeptidases:**

**trypsin**

**chymotrypsin**

**elastase**

## **Catalysis (ch 22)**

**definition of catalyst, energy diagrams illustrating catalytic action**

**types of catalysis:**

**acid: specific and general**

**base: specific and general**

**nucleophilic: stronger Nu are better catalysts**

**metal-ion: types of activation**

**intramolecular reactions**

**why are they faster? effective molarity etc...**

**intramolecular catalysis: examples of each class (acid, base, Nu, metal)**

**overview of enzyme catalyzed reactions:**

**names of enzyme (“substrate” .....ase)**

**lock-and-key model**

**induced-fit model**

**examples of enzyme catalysis**

**carboxypeptidase A, chymotrypsin: what is going on in the active site (what types of catalysis occur there?)**

## **Coenzymes (ch 23)**

**NAD<sup>+</sup>, NADP<sup>+</sup>, NADH, NADPH (structure, redox form, reactions)  
FMN, FAD, FADH<sub>2</sub> (structure, redox form, reactions)**

**lipoic acid (lipoate, redox forms)**

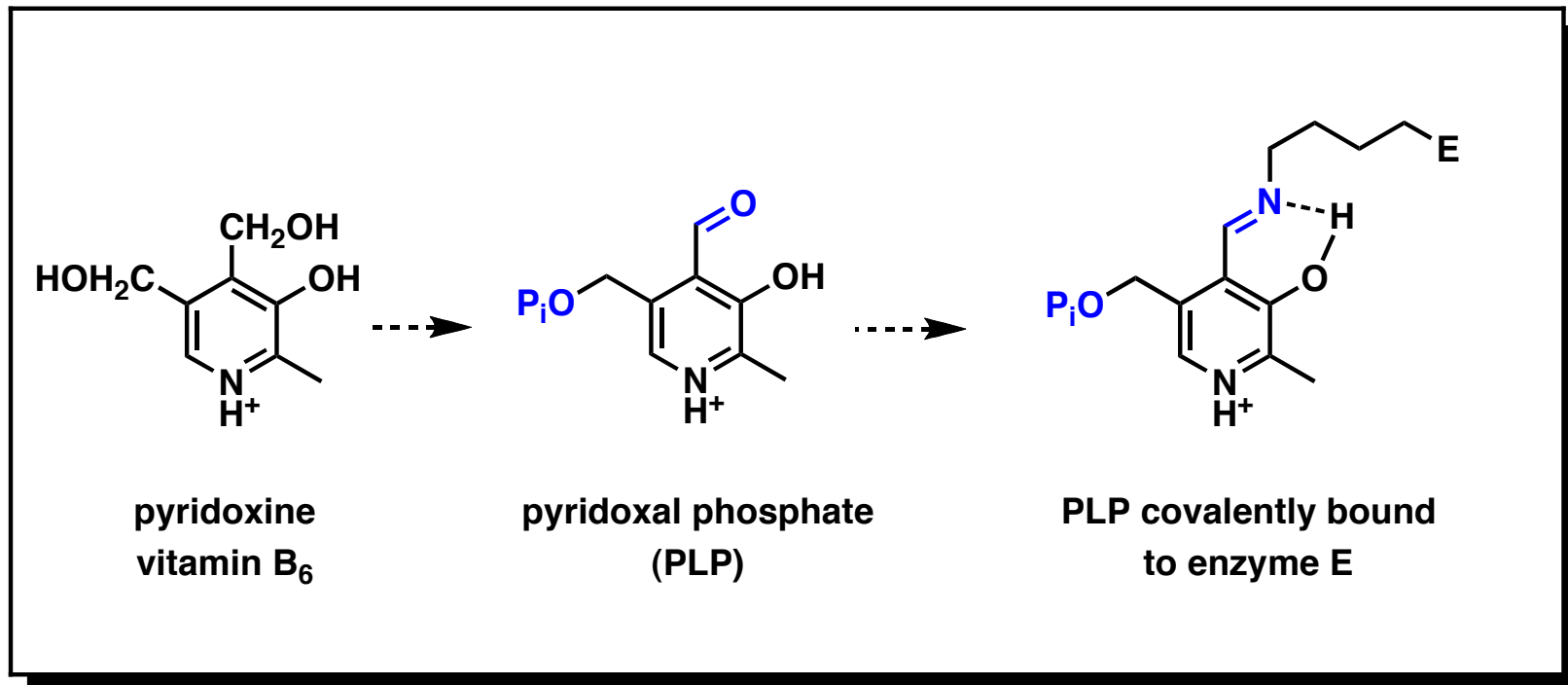
**TPP (thiamine pyrophosphate, structure, ylide form, type of reactions, PD complex)**

**biotin (structure, type of reaction, why ATP, why Mg<sup>2+</sup>?)**

**mechanisms**

# Pyridoxal (PLP)

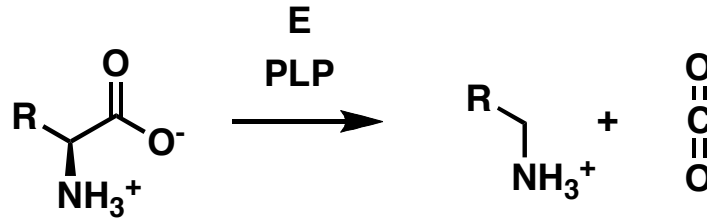
## PLP: from vitamin to active enzyme



## 5 reactions catalyzed by PLP

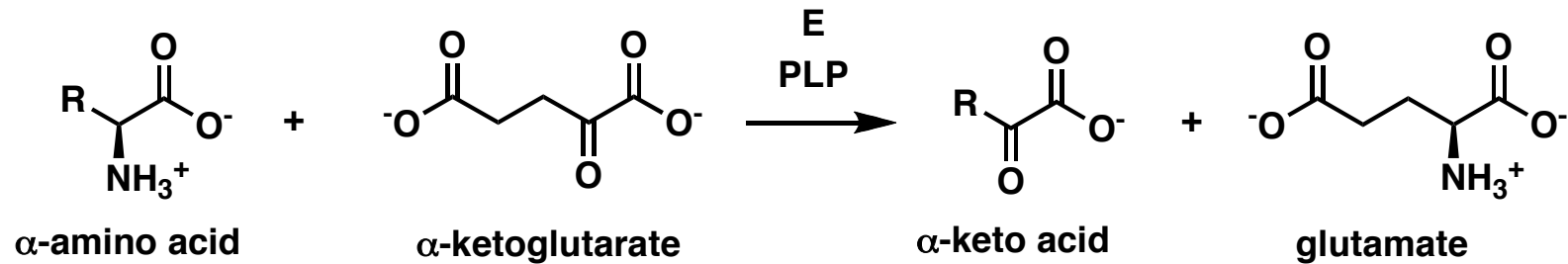
1

*decarboxylation*



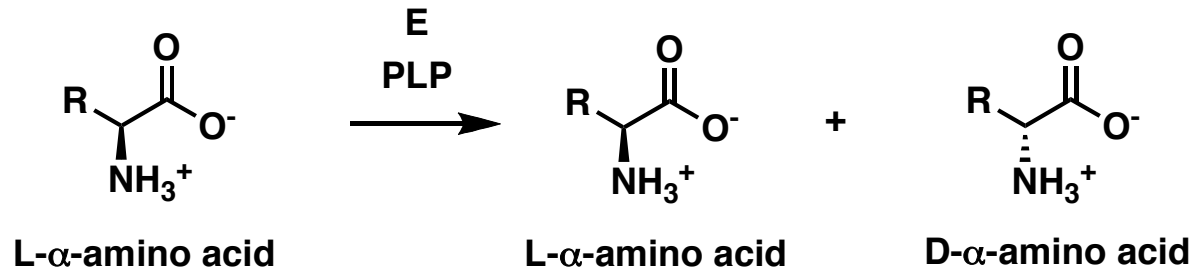
2

*transamination*

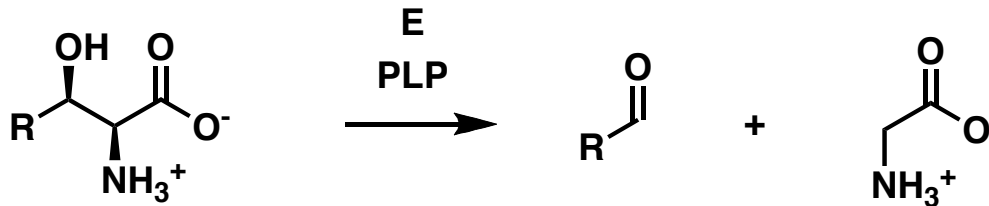


## 5 reactions catalyzed by PLP

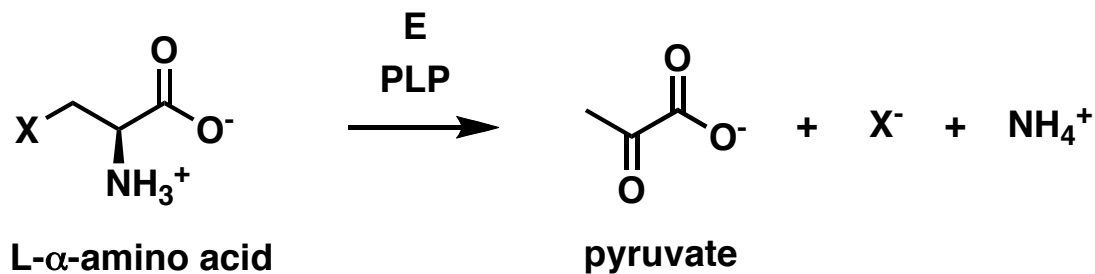
### 3 racemization



### 4 $C_{\alpha}$ - $C_{\beta}$ bond cleavage

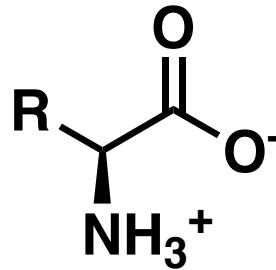


### 5 $\alpha,\beta$ -elimination



## common themes in all reactions with PLP

### 1. amino acid is the substrate



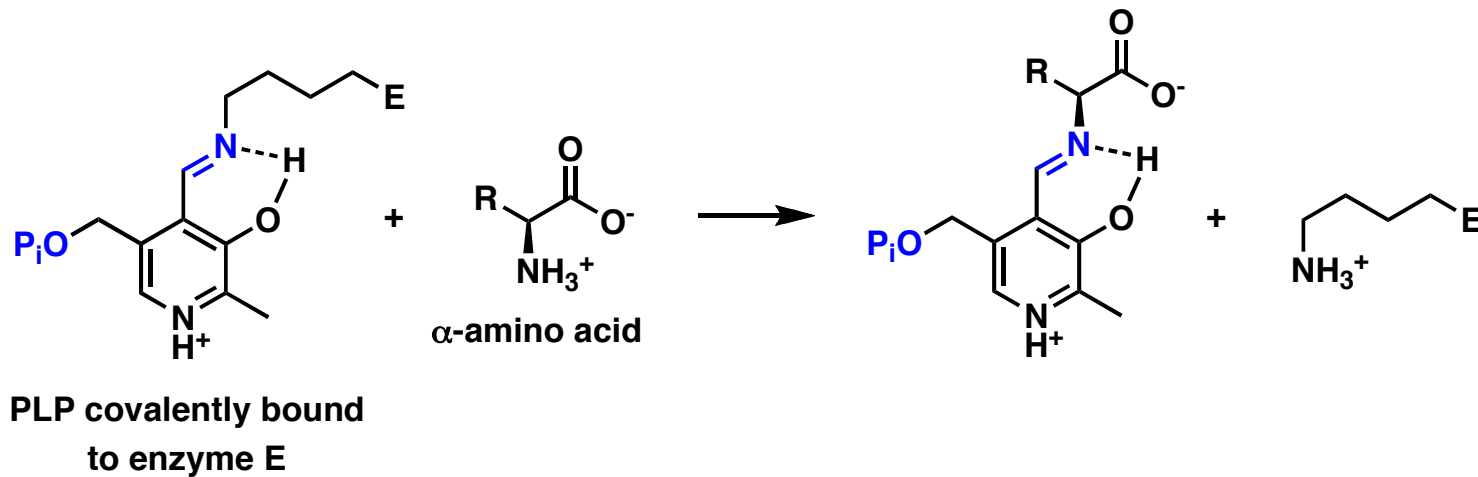
$\alpha$ -amino acid

all reactions take place at the  $\alpha$ -carbon, with different bonds involved



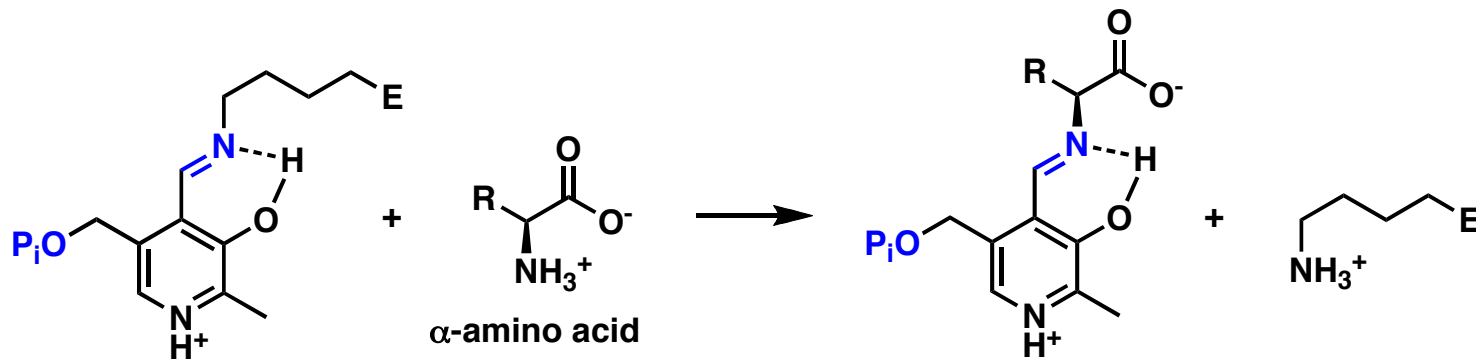
## common themes in all reactions with PLP

### 2. attachment of substrate by **transimination**:

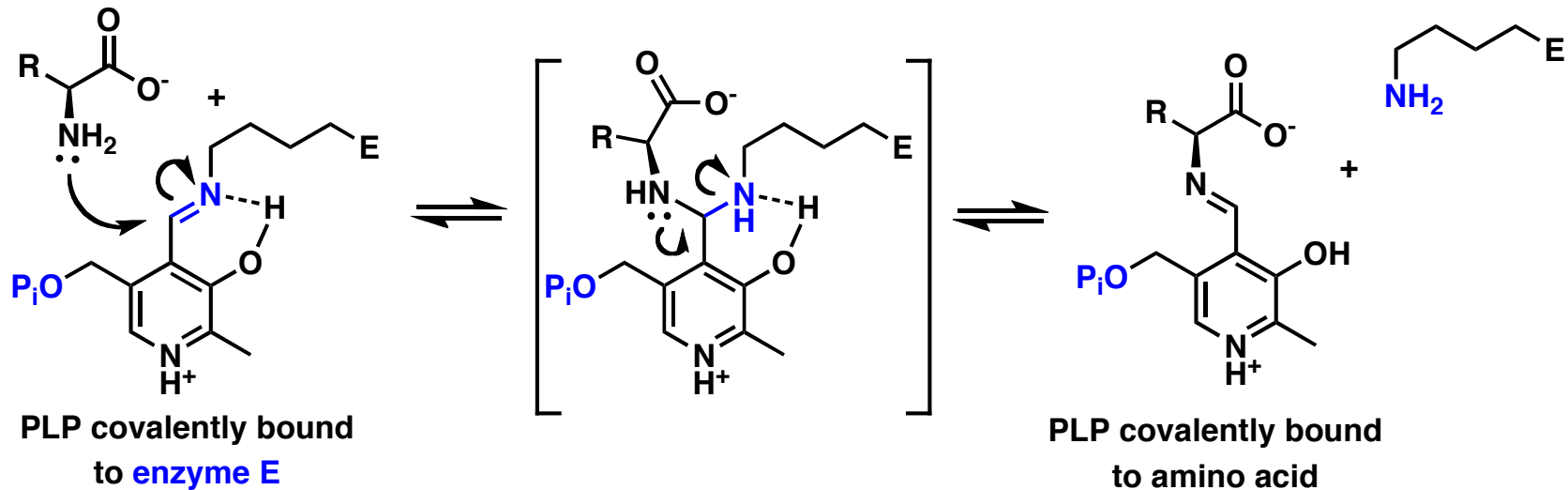


## common themes in all reactions with PLP

### 2. attachment of substrate by **transimination**:



PLP covalently bound  
to enzyme E

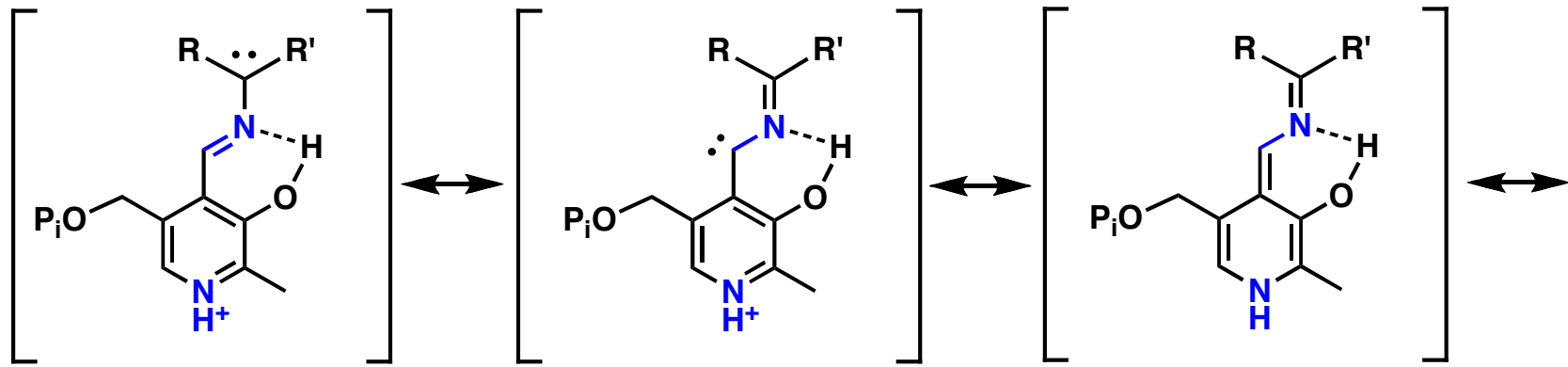
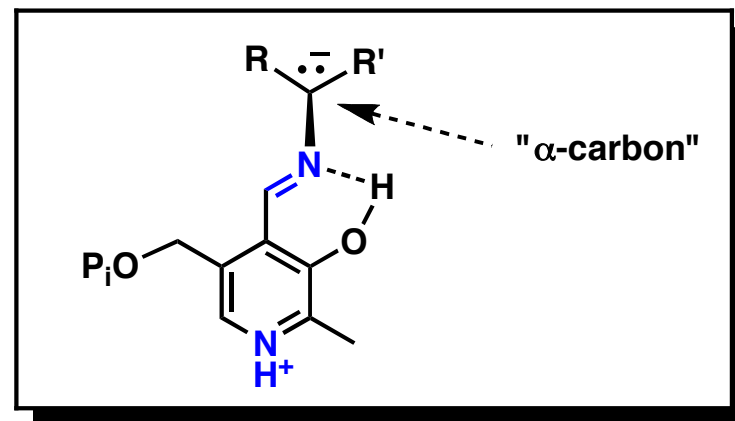


PLP covalently bound  
to enzyme E

PLP covalently bound  
to amino acid

## common themes in all reactions with PLP

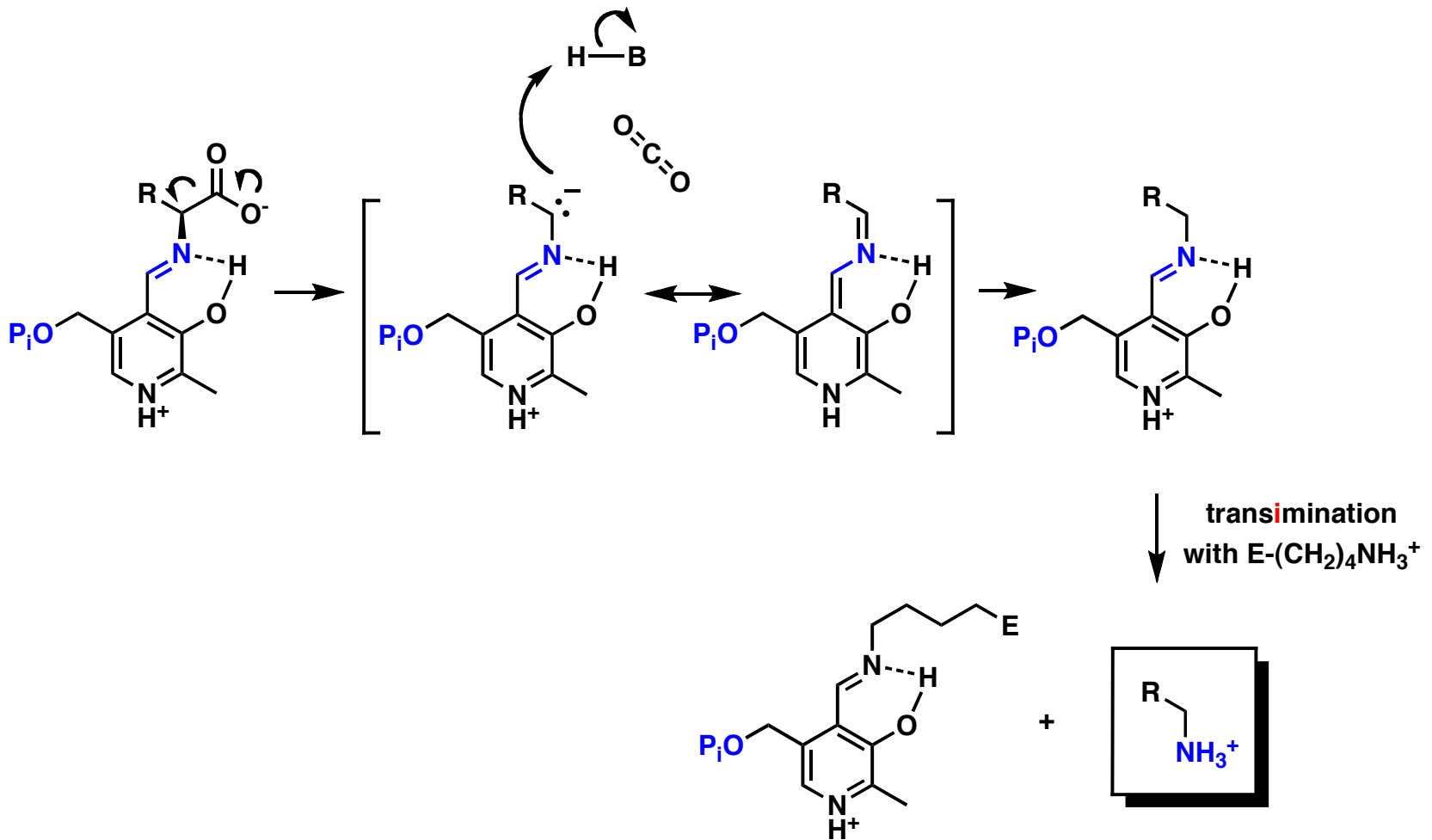
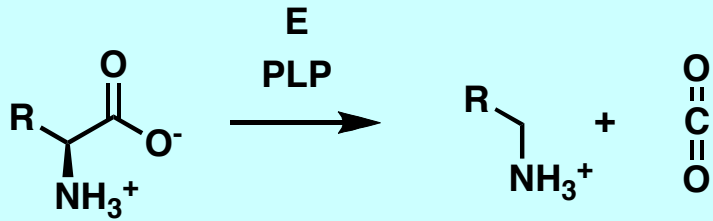
### 3. stabilization of negative charge at the $\alpha$ -carbon:



- stabilization through resonance/conjugation

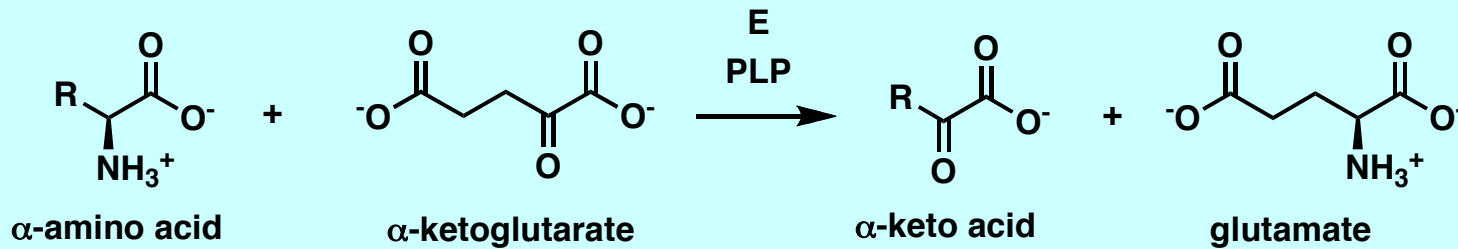
# Pyridoxal (PLP)

decarboxylation



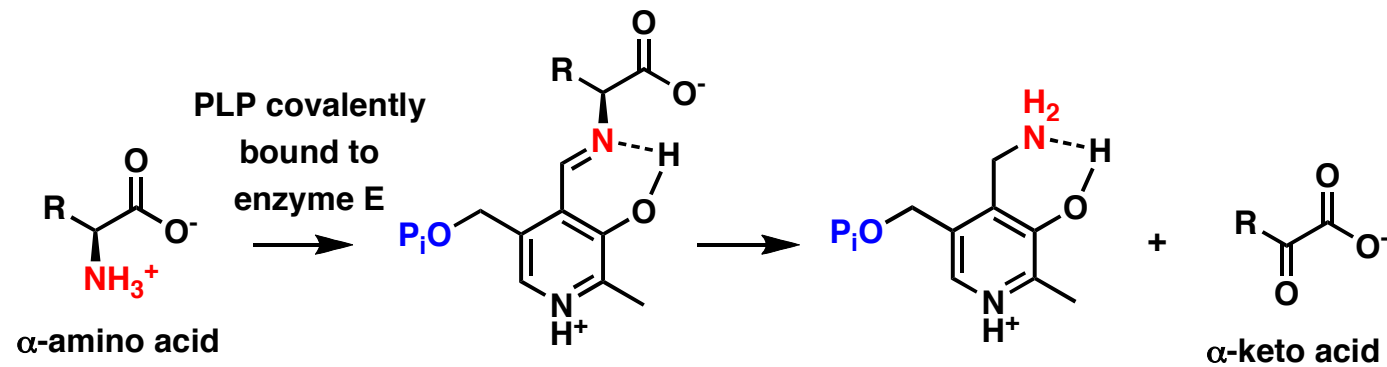
# Pyridoxal (PLP)

## transamination



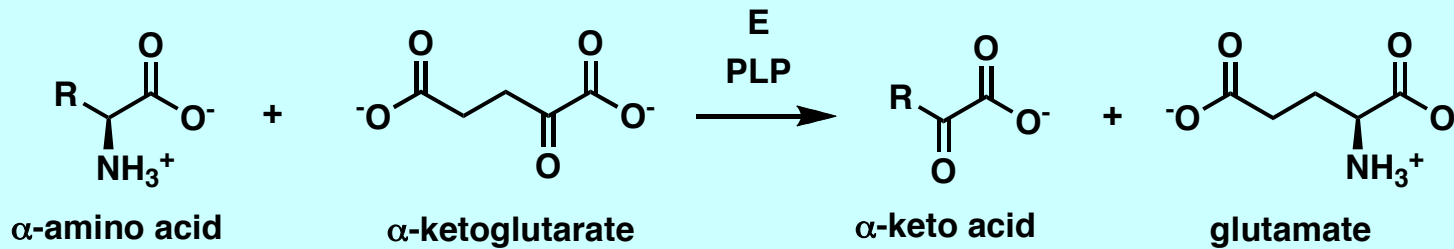
## summary of the mechanism:

### stage 1:



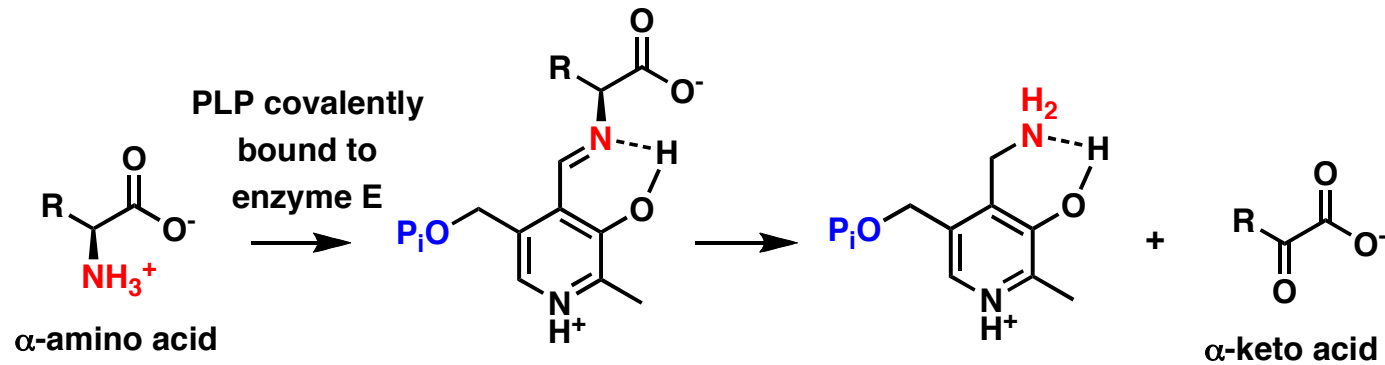
# Pyridoxal (PLP)

## transamination

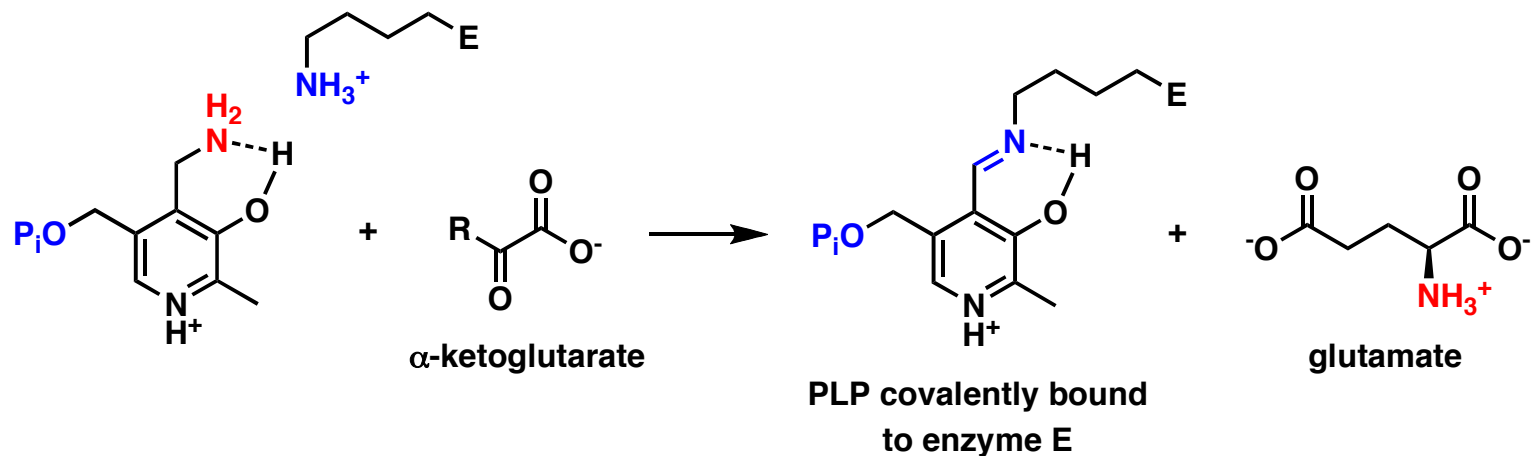


## summary of the mechanism:

### stage 1:

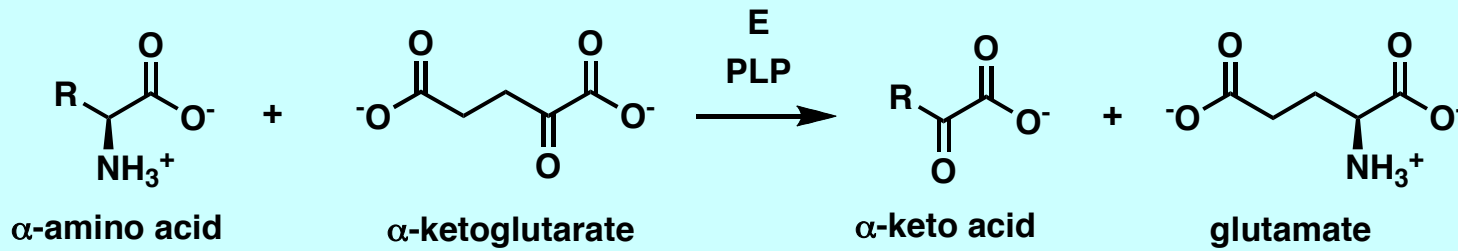


### stage 2:

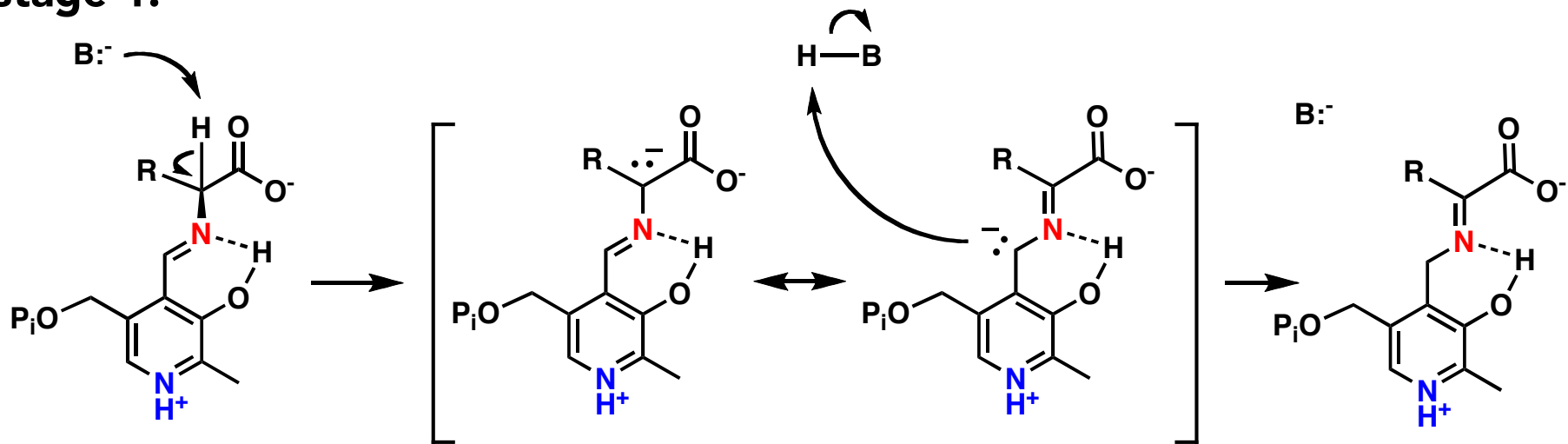


# Pyridoxal (PLP)

## transamination

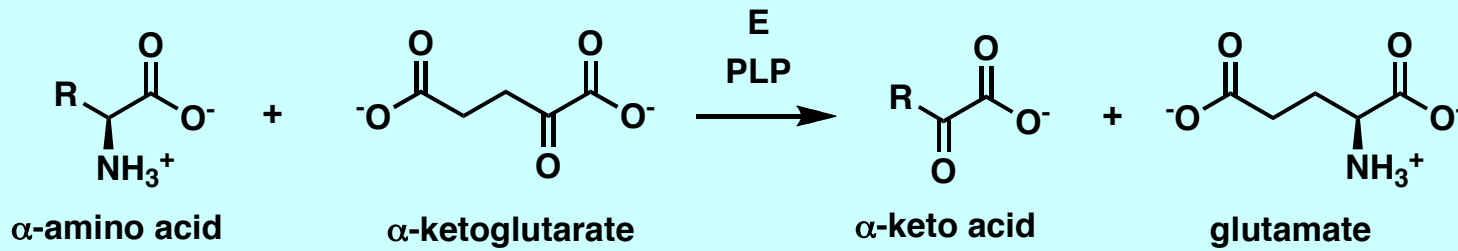


## stage 1:

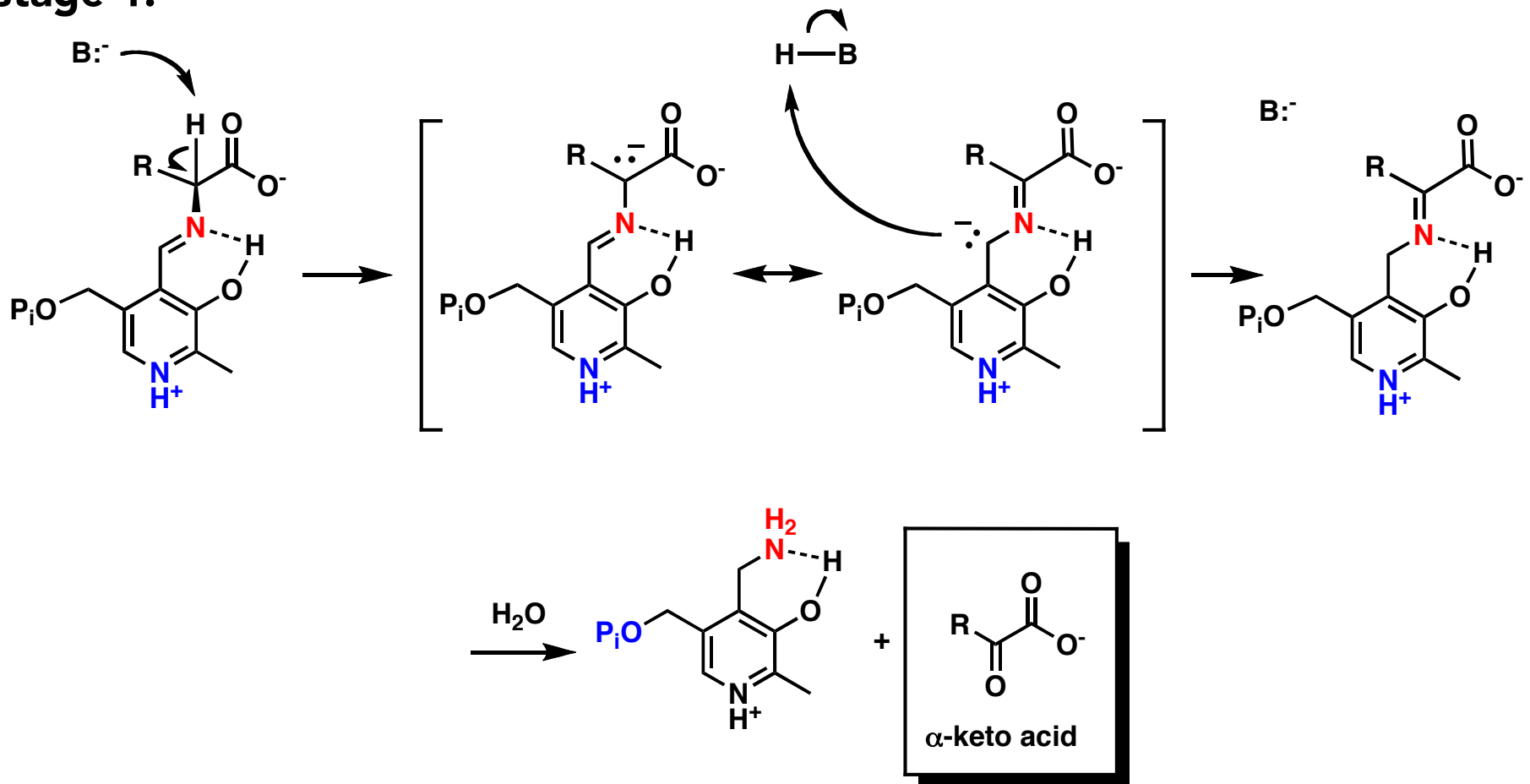


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## transamination



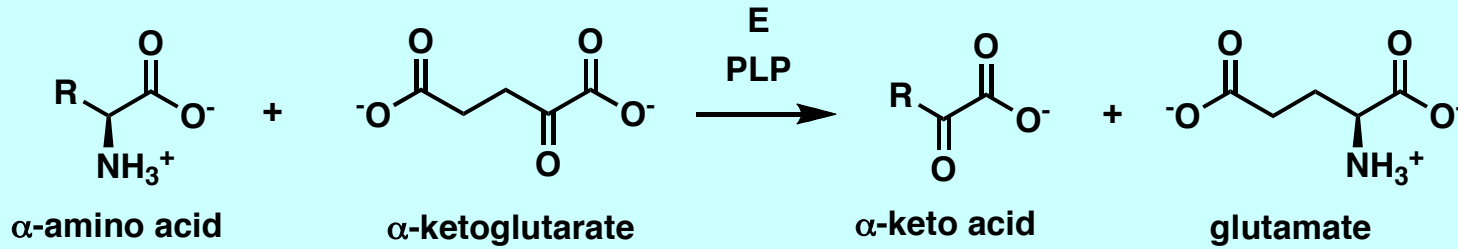
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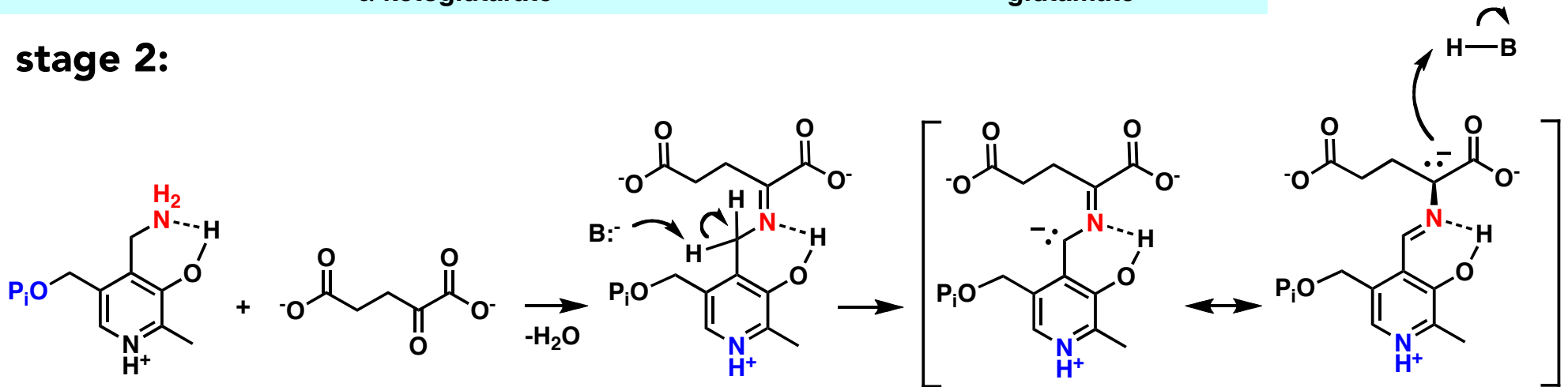


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## transamination

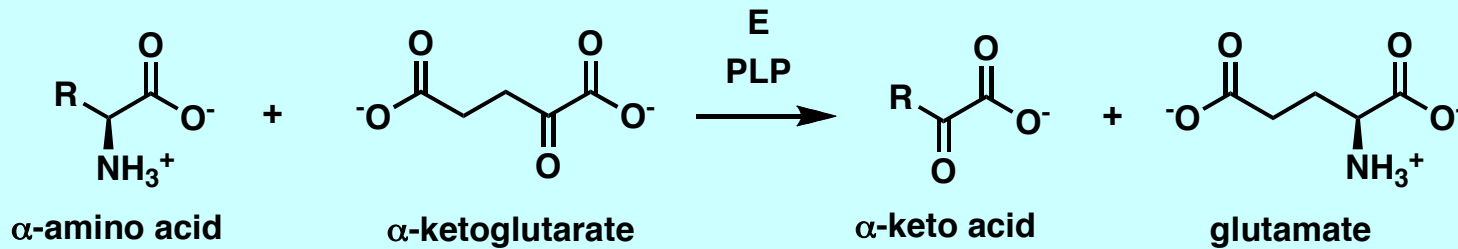


## stage 2:

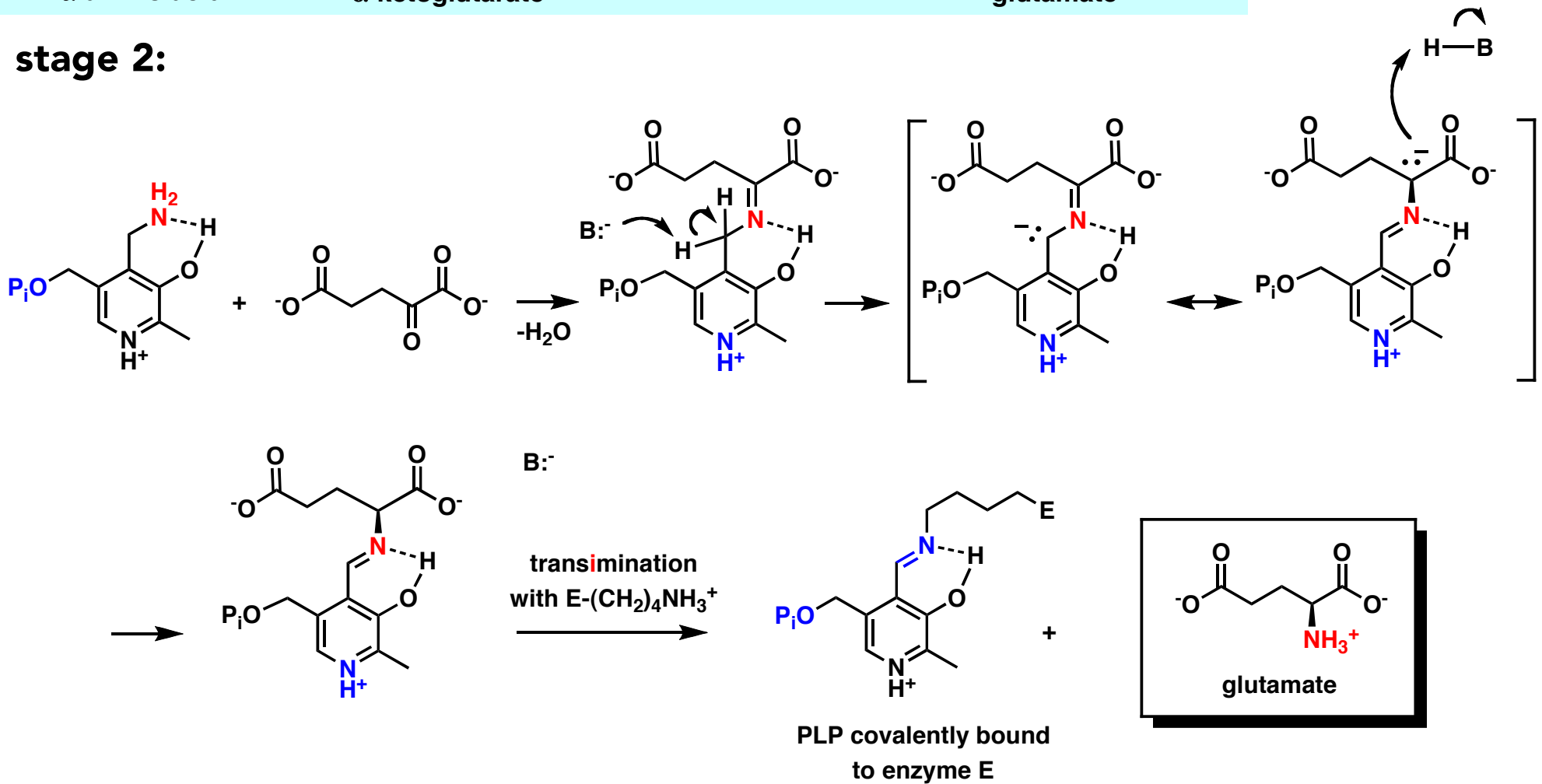


# Pyridoxal (PLP)

## transamination

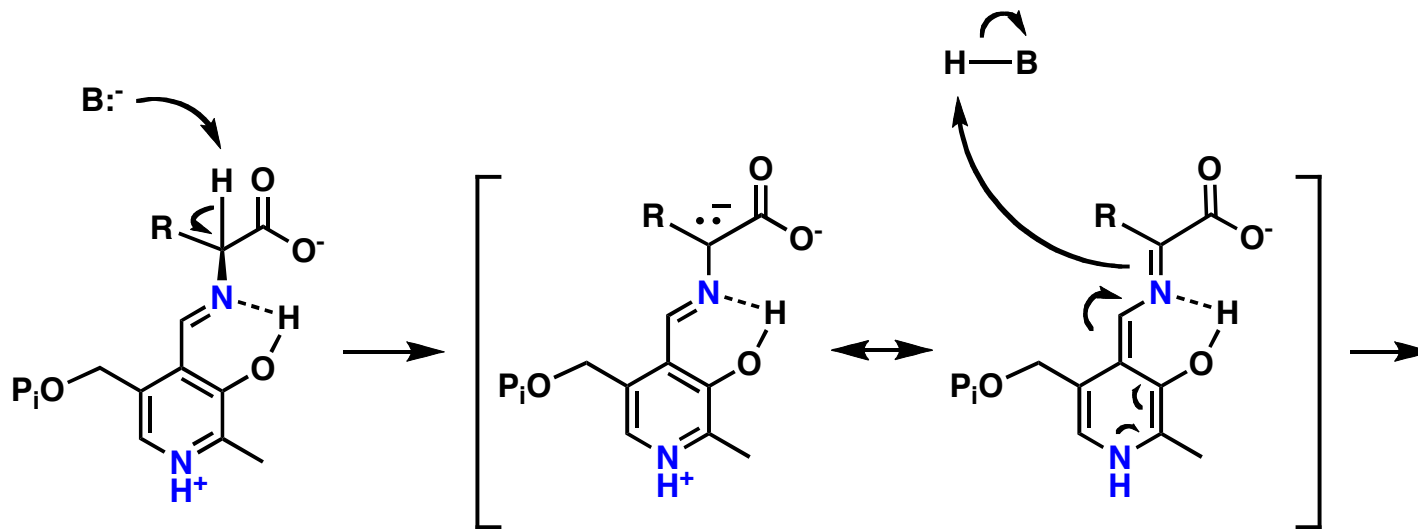
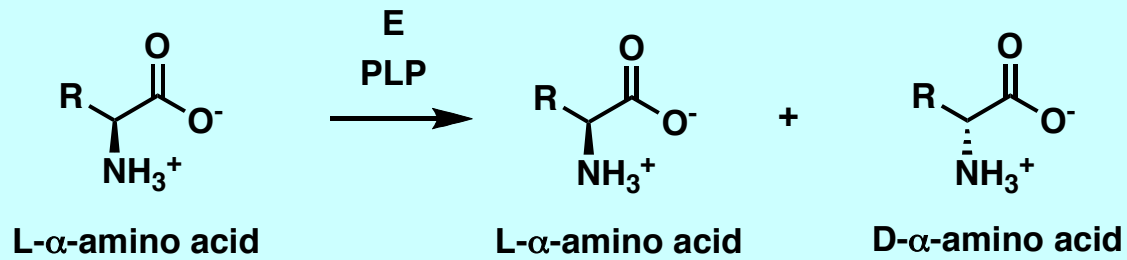


## stage 2:



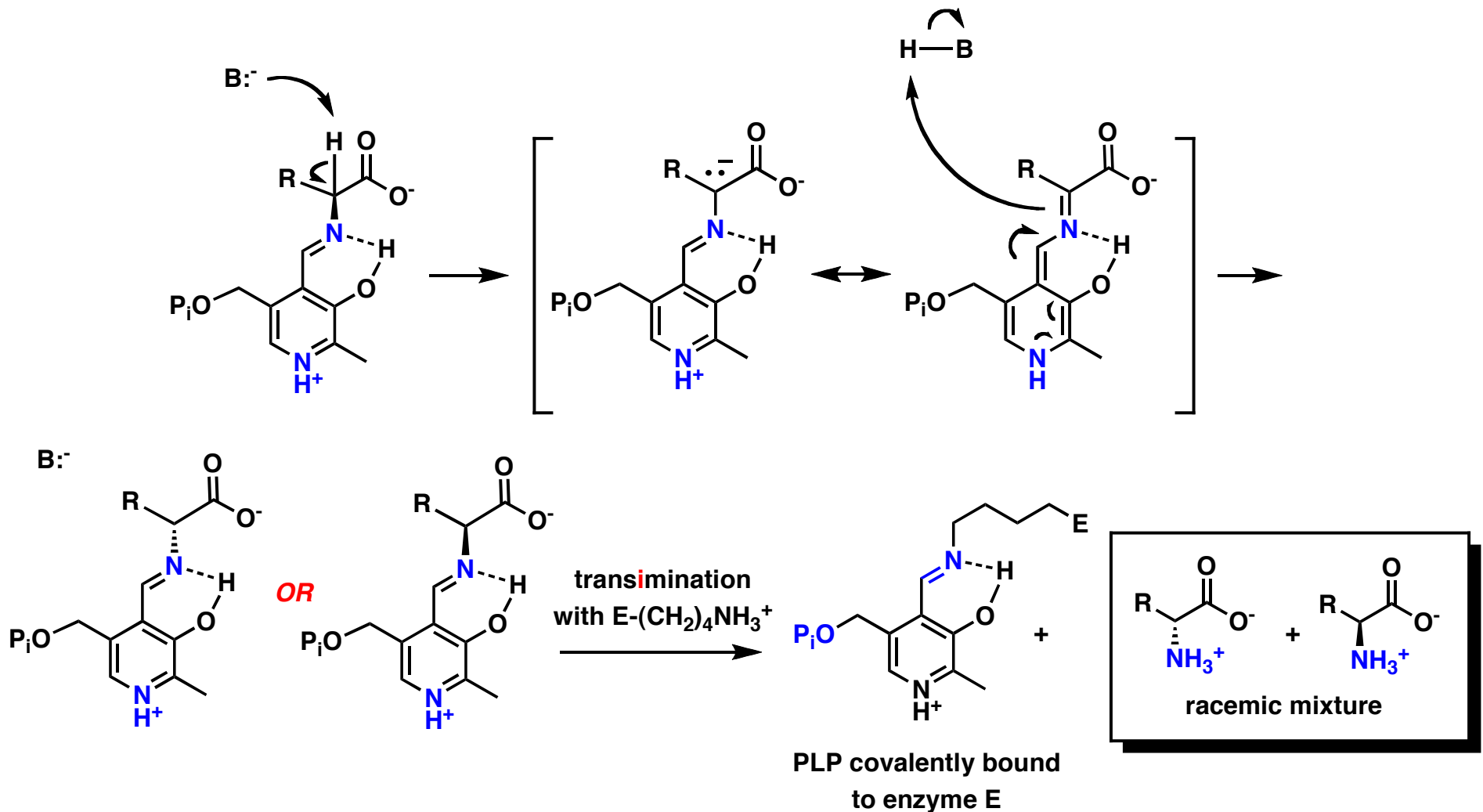
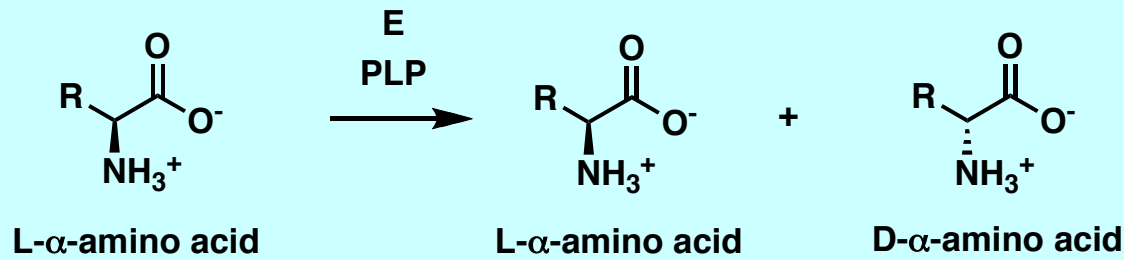
# Pyridoxal (PLP)

*racemization*



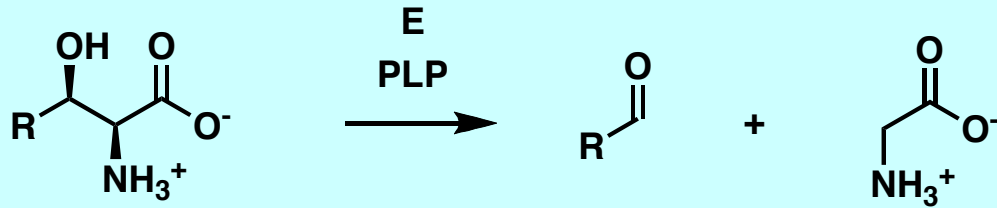
# Pyridoxal (PLP)

## racemization



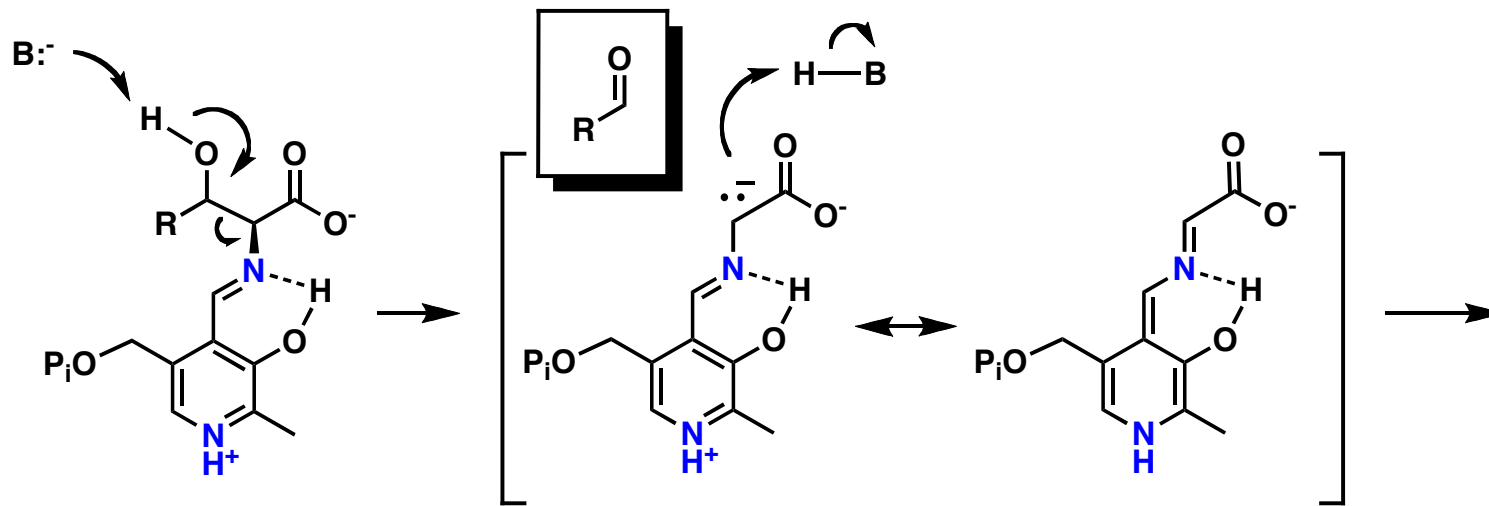
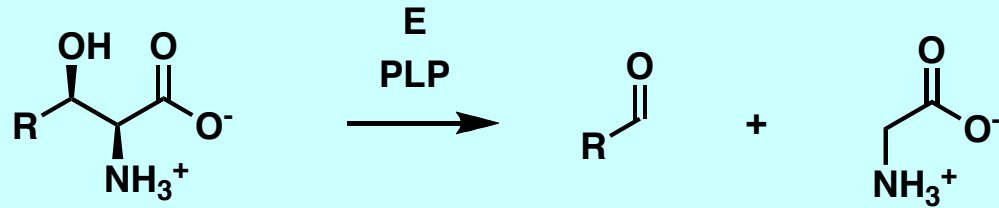
# Pyridoxal (PLP)

*C<sub>α</sub>-C<sub>β</sub> bond cleavage*



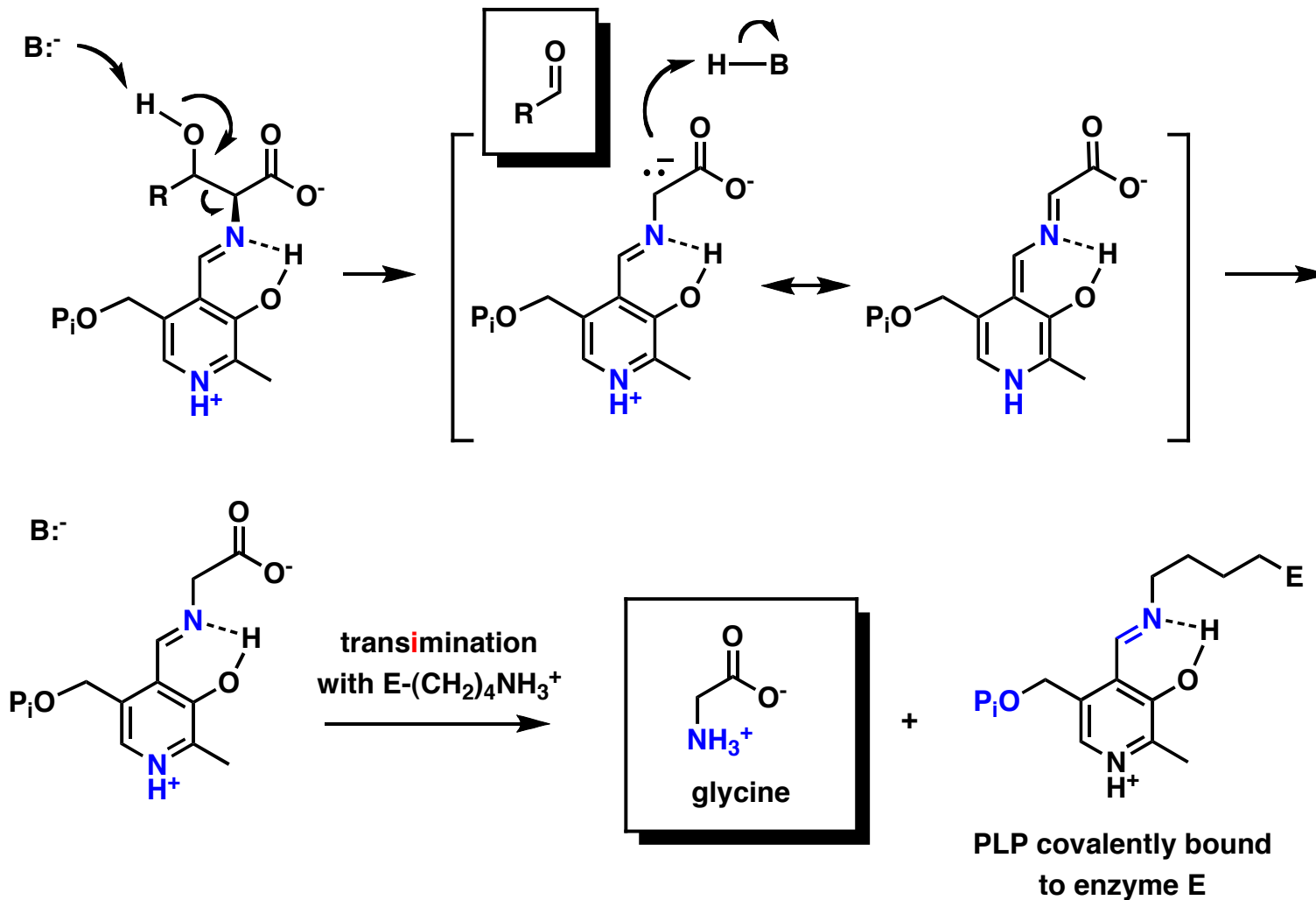
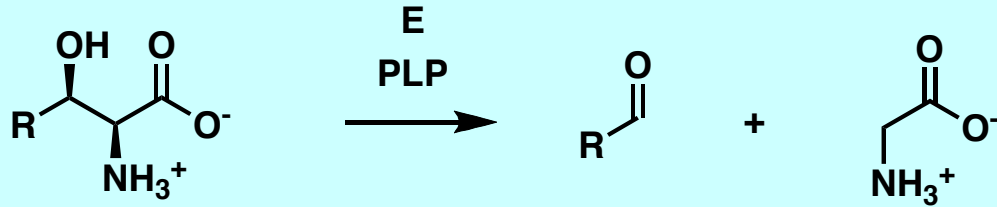
# Pyridoxal (PLP)

$C_{\alpha}$ - $C_{\beta}$  bond cleavage



# Pyridoxal (PLP)

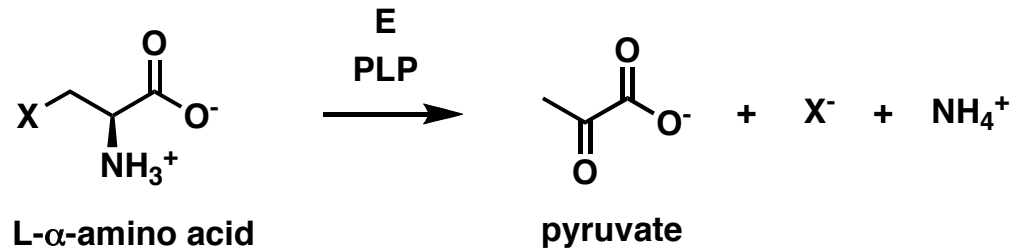
## $C_{\alpha}$ - $C_{\beta}$ bond cleavage



## PRACTICE PROBLEM 18

Propose a mechanism for PLP-catalyzed  $\alpha,\beta$ -elimination?

*$\alpha,\beta$ -elimination*



also, see related problem 34



## PRACTICE PROBLEM 15, 16

**Explain why the ability of PLP to catalyze its reactions is greatly reduced if they are carried out at a pH at which the pyridine nitrogen is not protonated.**

**Explain why the ability of PLP to catalyze an amino acid transformation is greatly reduced if the OH group of pyridoxal phosphate is replaced by OCH<sub>3</sub>?**