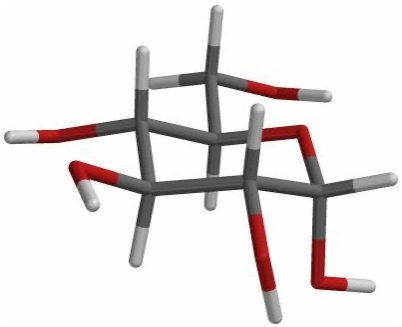

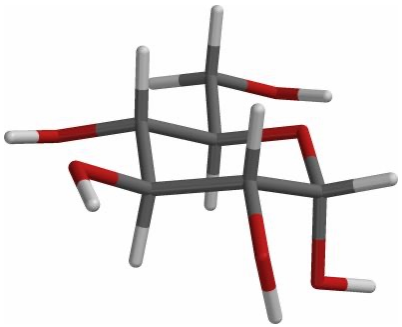

Chem 109 C
Bioorganic Compounds

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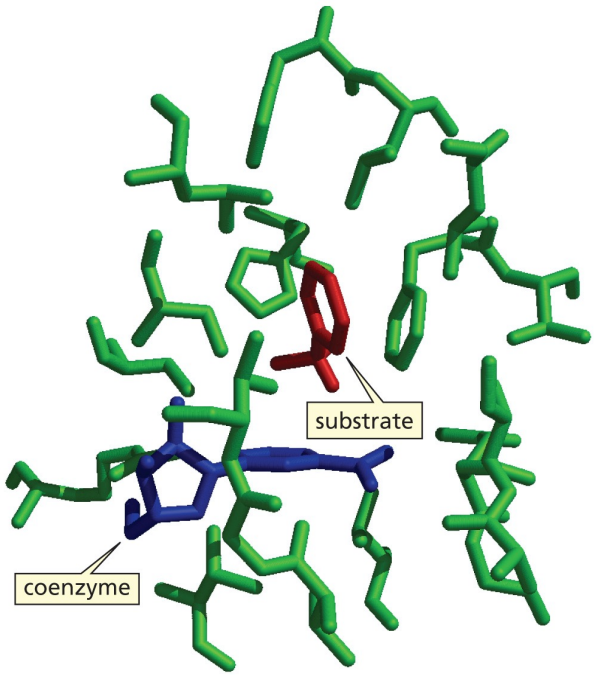
Chapter 23: Coenzymes

Overview and Introduction:

- enzymes are biological catalysts
- many enzymes are inactive without cofactors
- cofactors are 1) metal ions or 2) coenzymes
- coenzymes - organic molecules, derived from vitamins



Chapter 23: Coenzymes

Coenzyme	Vitamin	Reaction catalyzed
NAD^+, $NADP^+$ / NADH, NADPH	niacin, nicotinamide	oxidation/reduction of alcohols
FAD / $FADH_2$	riboflavin (B2)	oxidation/reduction, other
Thiamine pyrophosphate TPP	thiamine (B1)	acyl group transfer
Lipoic acid /dihydropipoic acid	lipoic acid	oxidation/reduction
Coenzyme A, CoASH	pantothenic acid (B5)	acyl group transfer
Biotin	biotin (B7)	carboxylation
Pyridoxal phosphate PLP	pyridoxin (B6)	6 amino acid reactions
Coenzyme B_{12}	vitamin B12	isomerization
Tetrahydrofolic acid, THF	folic acid	one-carbon transfer
Vitamin KH_2	vitamin K	carboxylation

Vitamin KH_2 is not soluble in water

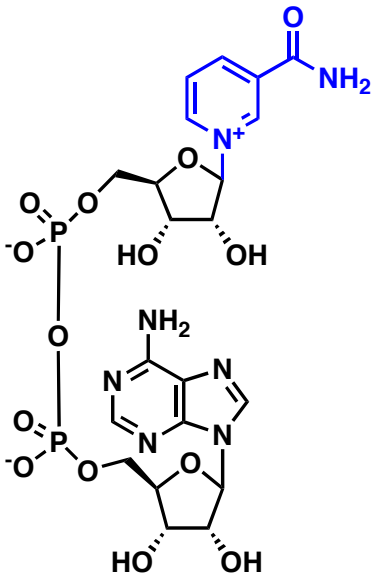
see Table 23.1 in Chapter 23

Chapter 23: Coenzymes

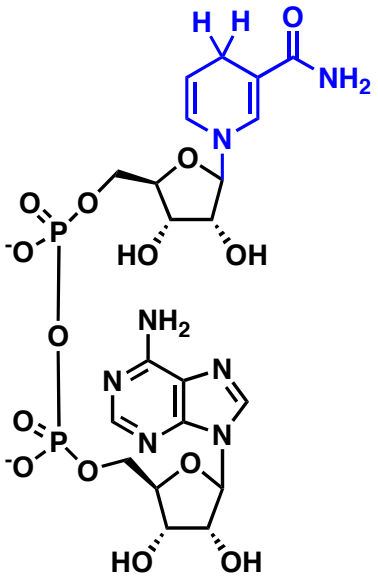
factors characterizing a coenzyme:

- **chemical structure**
- **associated vitamin**
- **type of reaction catalyzed**
- **reaction mechanism**
- **dietary source**
- **associated disease**

NAD^+-NADH, $NADP^+$-NADPH



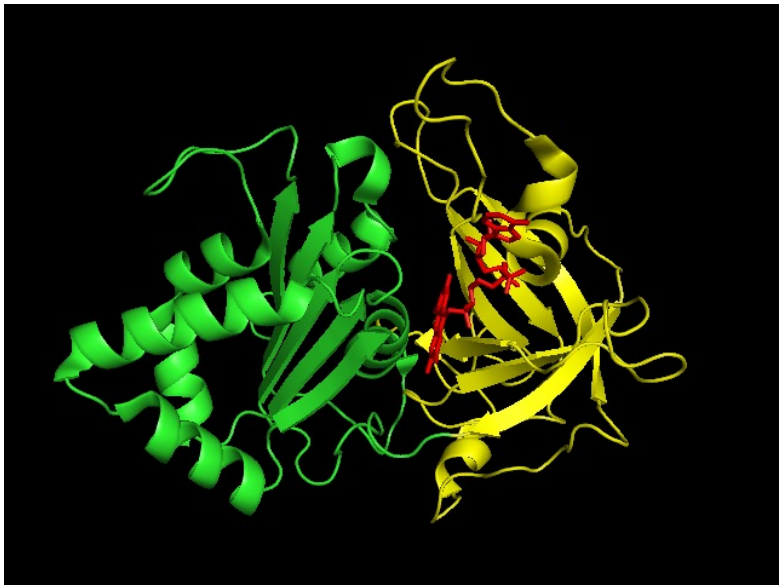
NAD^+



NADH

vitamin:

nicotinamide

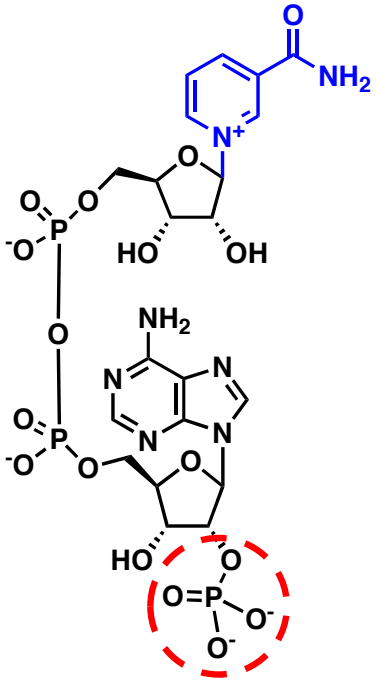


- NAD^+ is a catabolic enzyme
- $[NAD^+]/[NADH]$ ~ 1000 : 1 (cytosol)

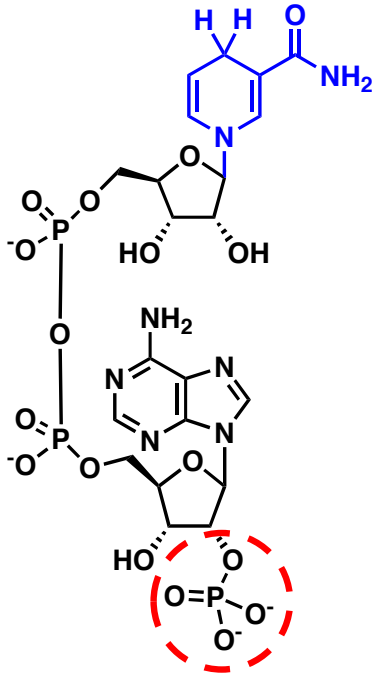
  [0.3 mM]

- catalyze redox (oxidation-reduction) reactions
- source of nicotinamide: meats, vegetables, peanuts etc.
- deficiency disease: pellagra (skin lesions, sensitivity to light etc.)

NAD^+-NADH, $NADP^+$-NADPH



$NADP^+$



NADPH

vitamin:

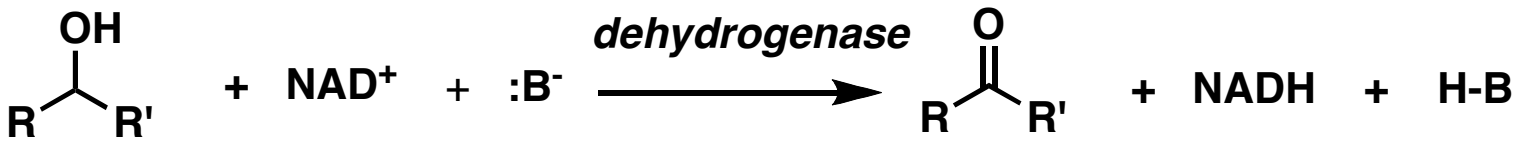
nicotinamide

- $NADP^+$ is an anabolic enzyme
- $[NADP^+]/[NADPH]$ ~ 1 : 100

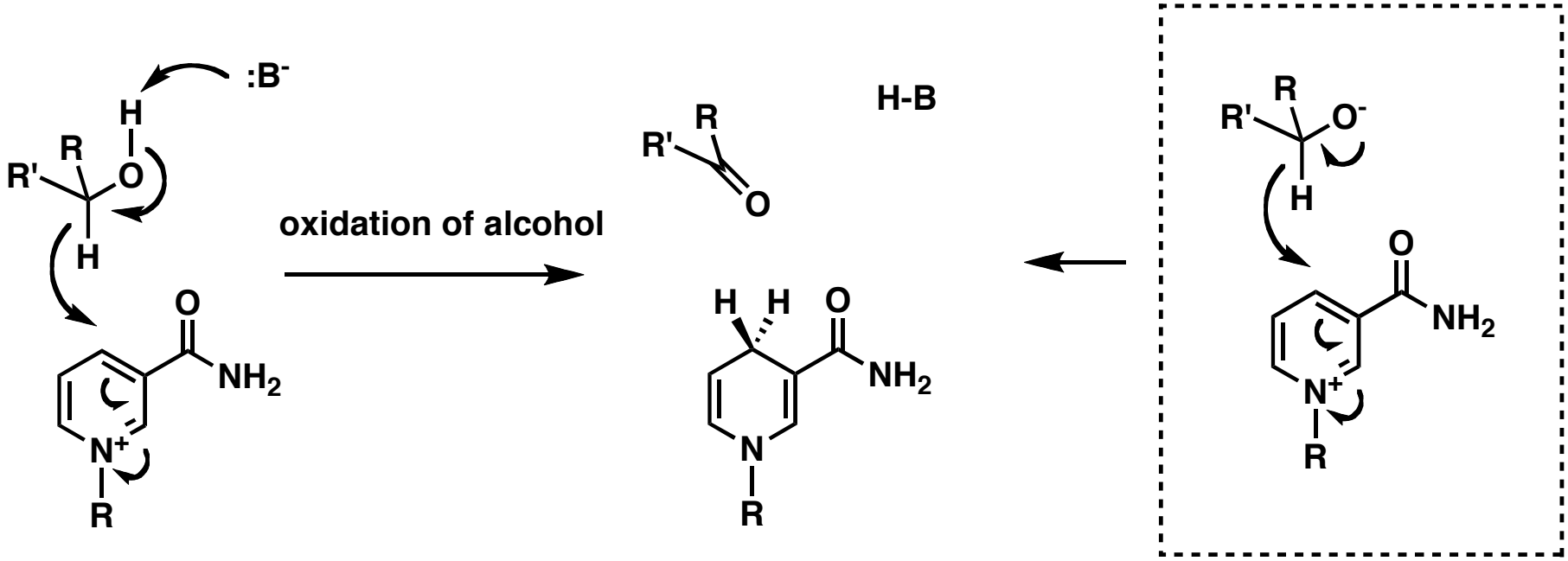
- catalyze redox (oxidation-reduction) reactions
- source of nicotinamide: meats, vegetables, peanuts etc.
- deficiency disease: pellagra (skin lesions, sensitivity to light etc.)

NAD$^+$, NADP$^+$

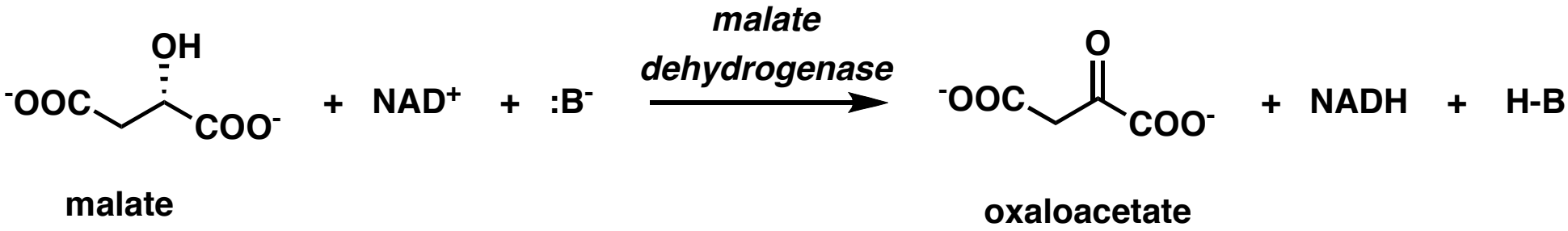
Oxidation with NAD$^+$ (or NADP$^+$):



General mechanism of oxidation:

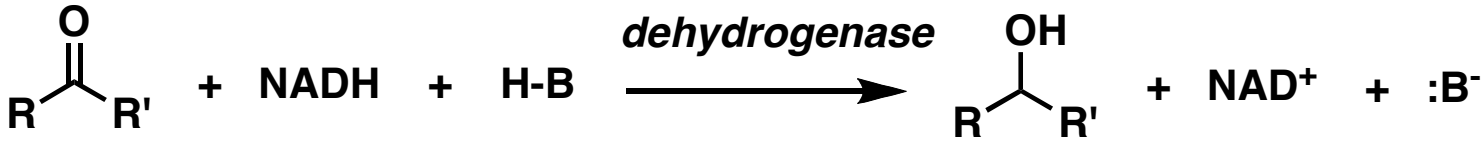


Oxidation with NAD+ (or NADP+), examples:

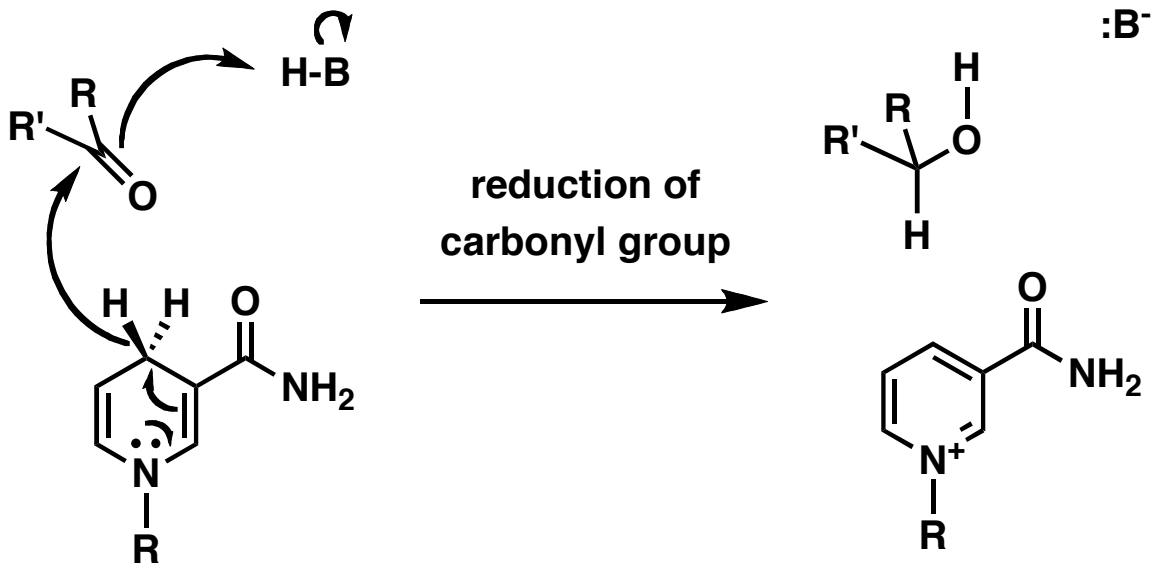


- important reaction in the citric acid cycle

Reduction with NADH (or NADPH), a reverse process:

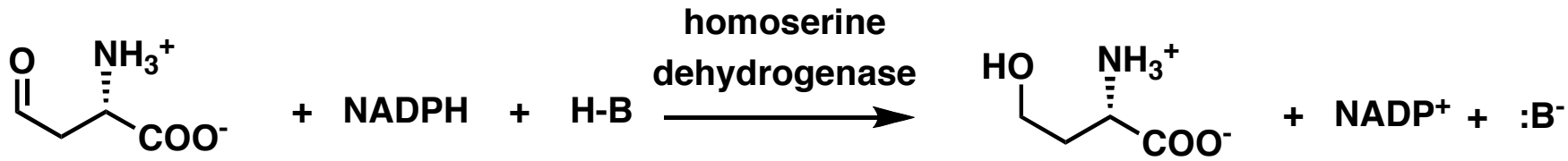


General mechanism of reduction:



- **NADH and NADPH are H^- donors**

Reduction with NADPH (or NADH), examples:

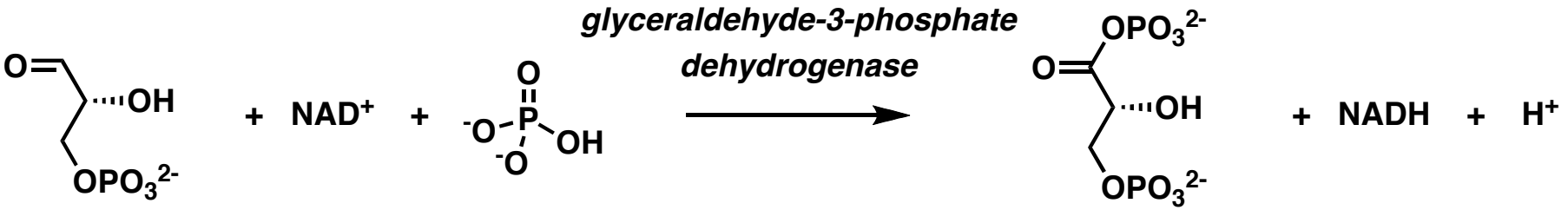


β-aspartate semialdehyde

homoserine

- important reaction in an anabolic pathway

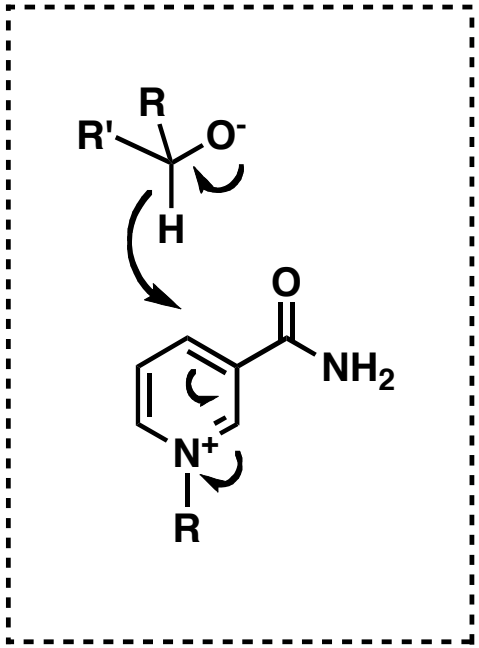
Oxidation - a more complex example:



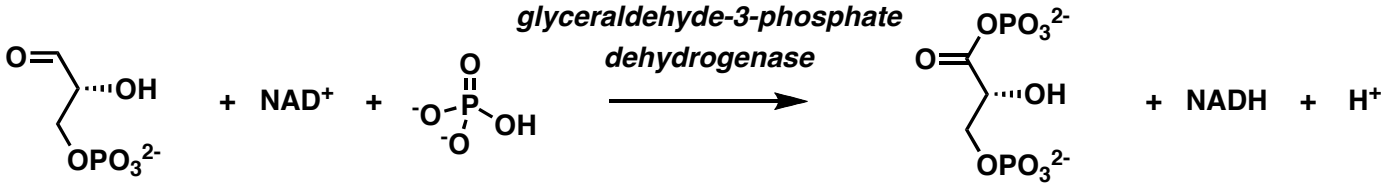
D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

G3P

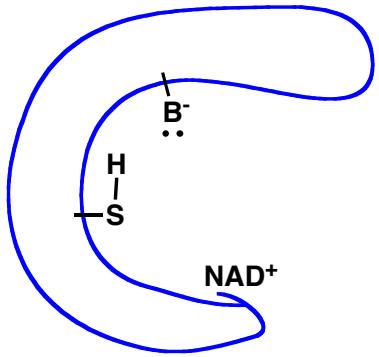


NAD^+, $NADP^+$

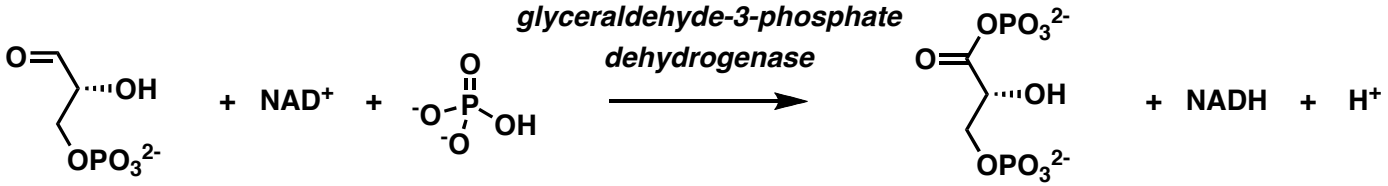


D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

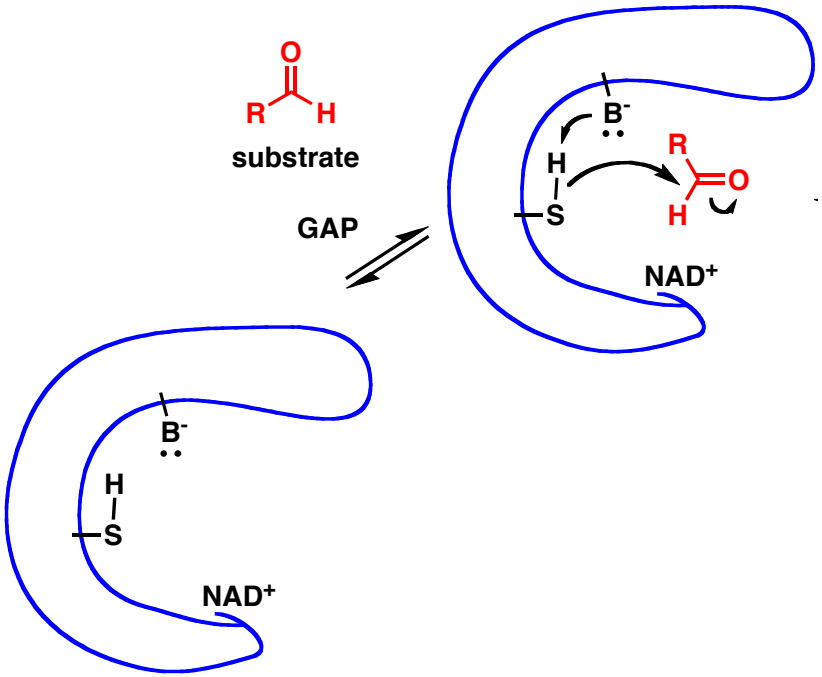


NAD^+, $NADP^+$

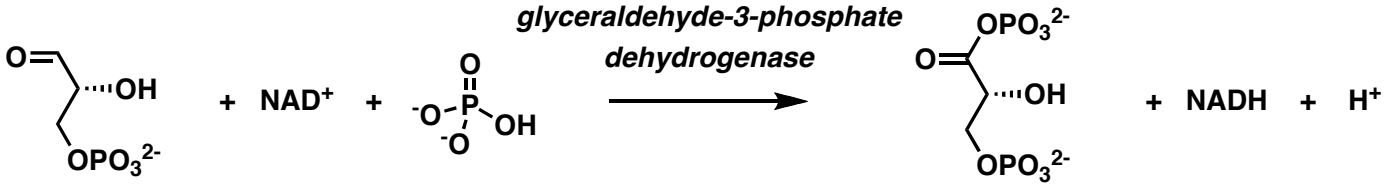


D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

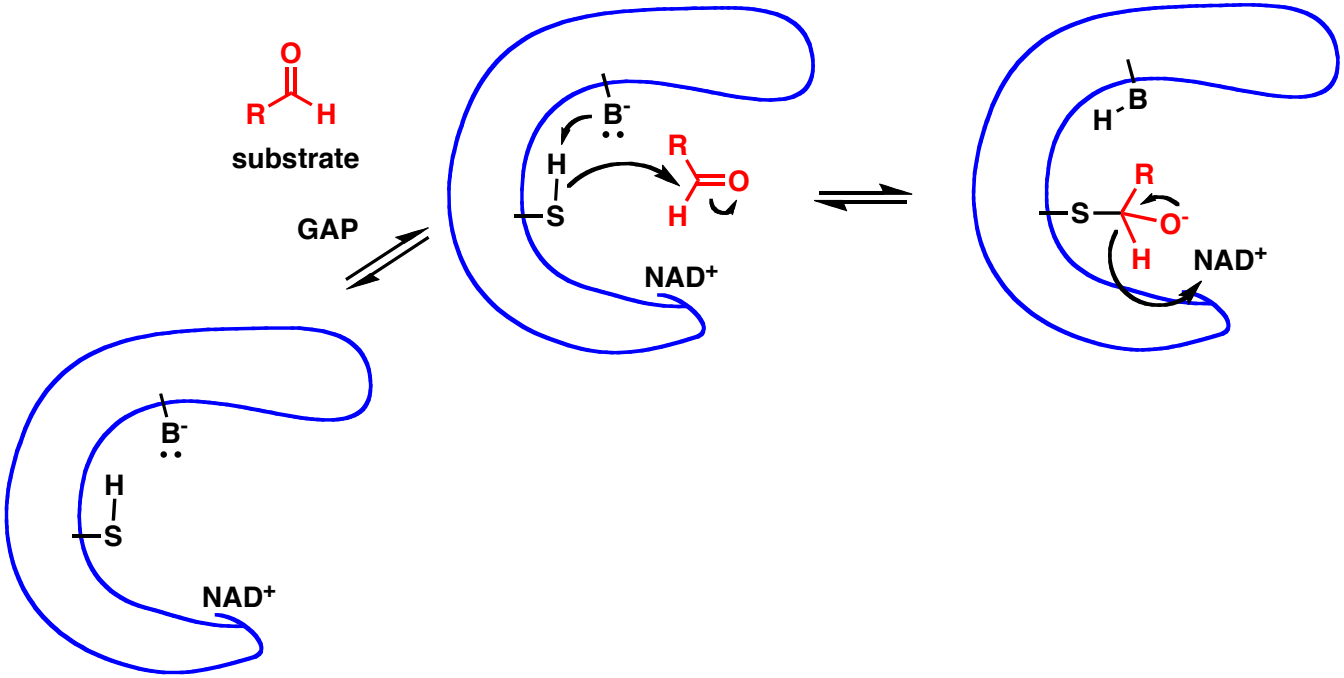


NAD+, NADP+

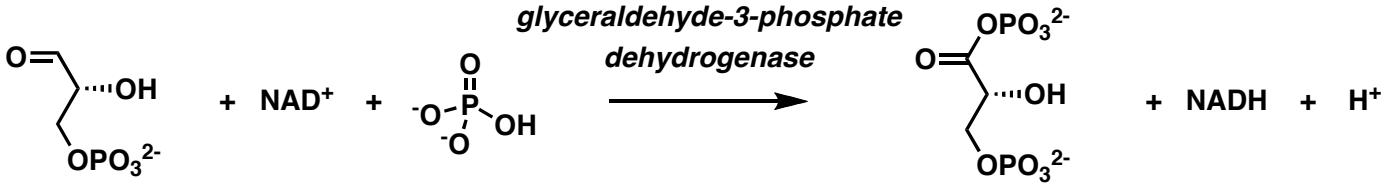


D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

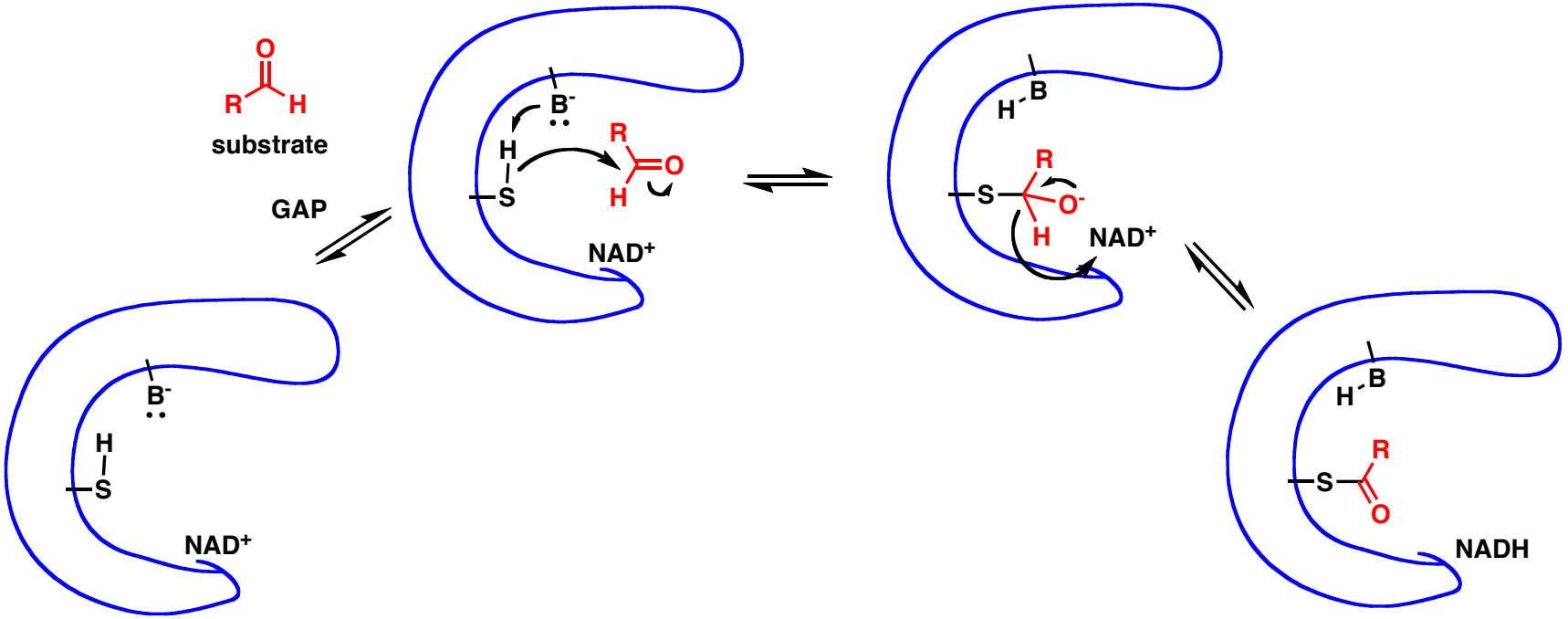


NAD^+, $NADP^+$

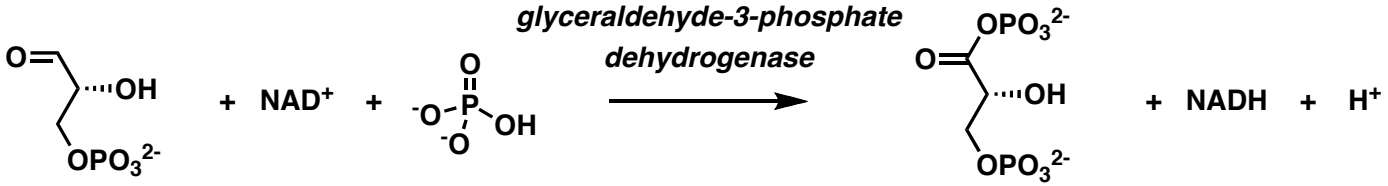


D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

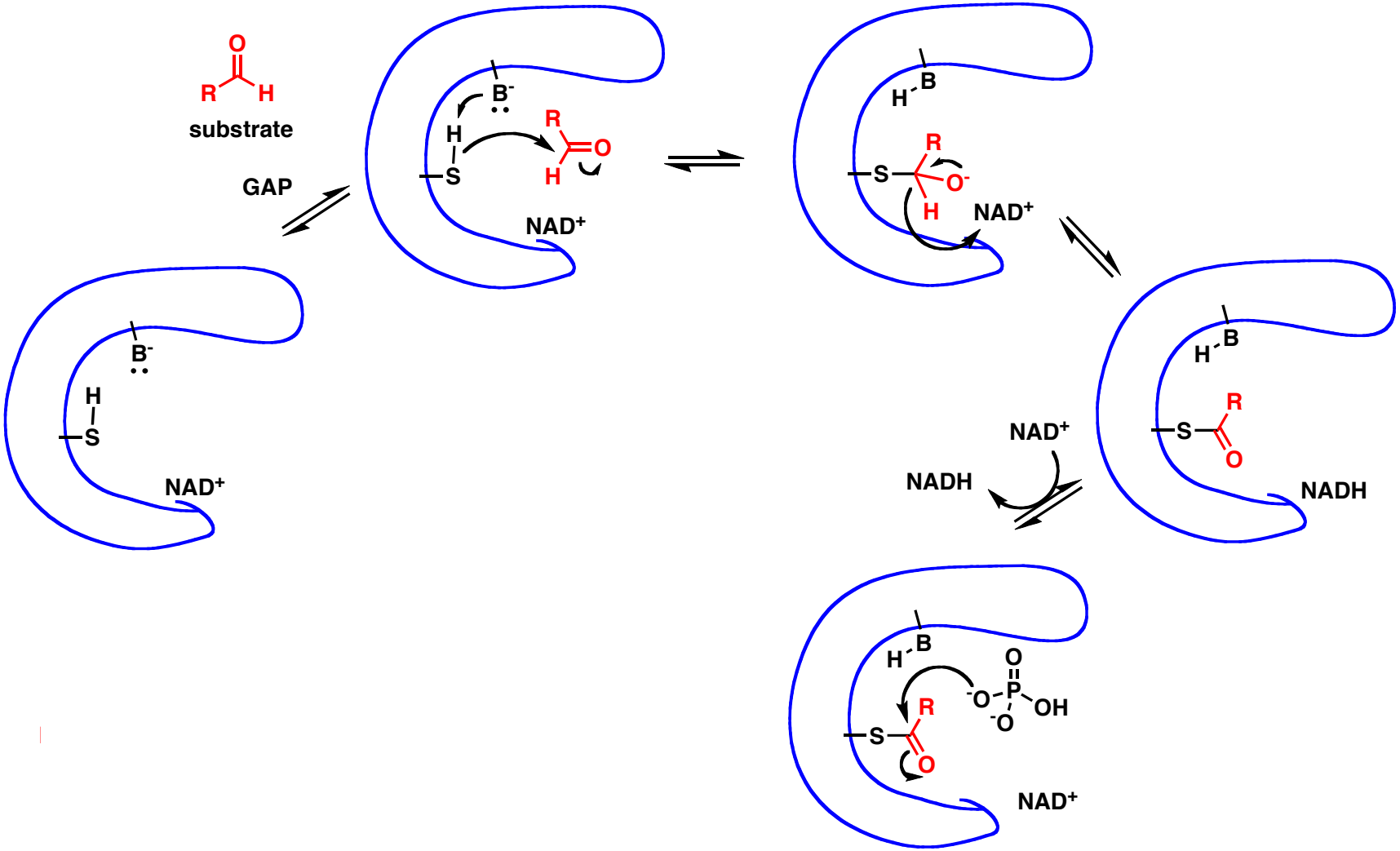


NAD+, NADP+

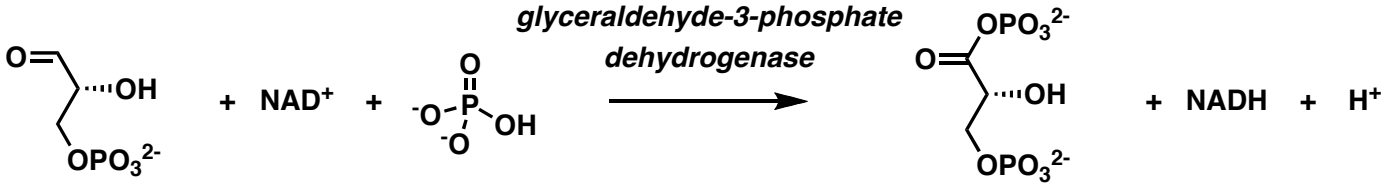


D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate

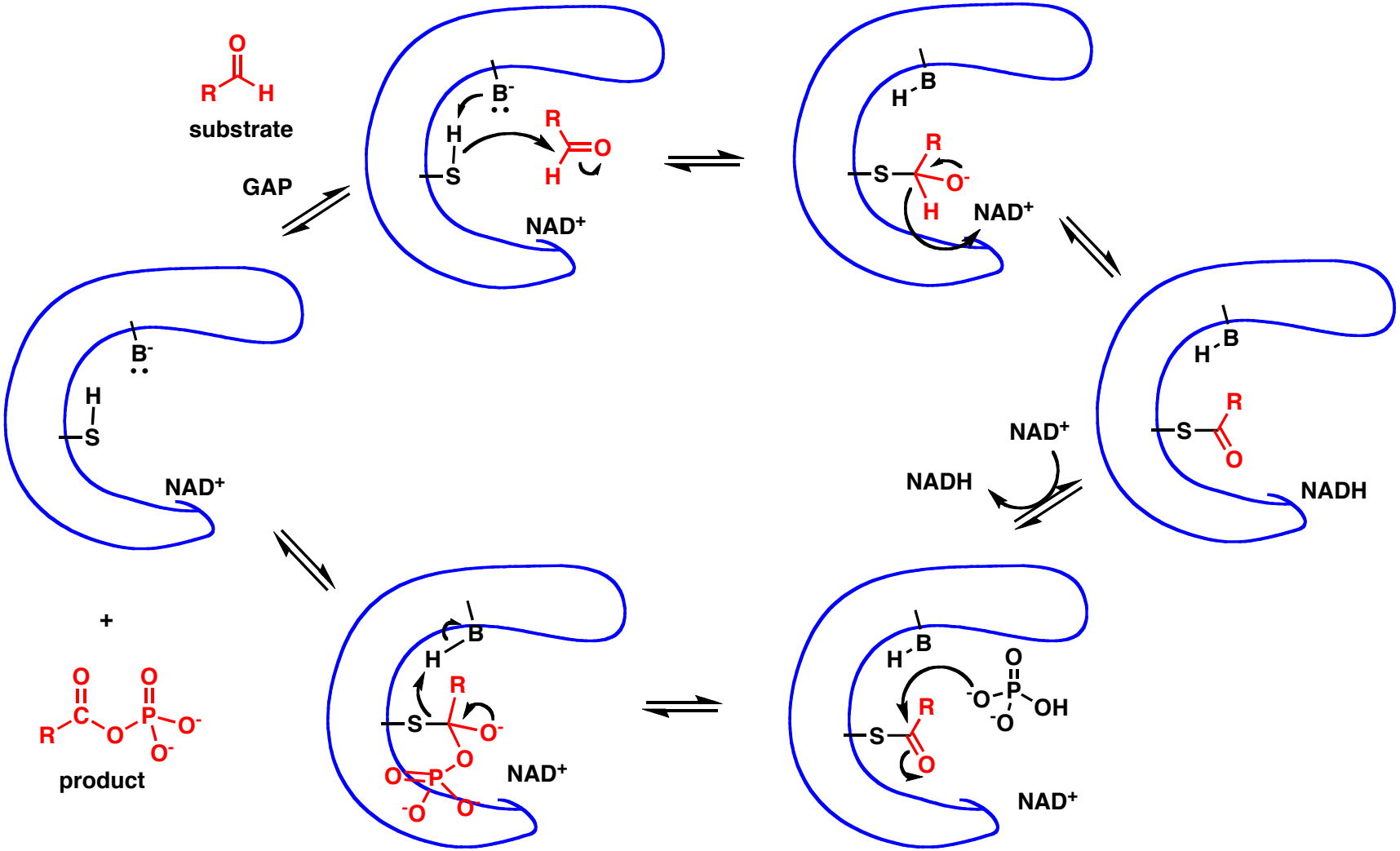


NAD+, NADP+



D-glyceraldehyde-3-phosphate

D-1,3-diphosphoglycerate



Stereochemistry of enzymatic reactions: enzyme reactions are stereospecific